

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
20 December 2001 (20.12.2001)

PCT

(10) International Publication Number  
**WO 01/96467 A2**

(51) International Patent Classification<sup>7</sup>: **C08L 51/00**

(21) International Application Number: PCT/IB01/01022

(22) International Filing Date: 8 June 2001 (08.06.2001)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:  
09/596,092 16 June 2000 (16.06.2000) US

(71) Applicant: **BASELL TECHNOLOGY COMPANY**  
B.V. [NL/NL]; Hocksteen 66, NL-2132 MS Hoofddorp  
(NL).

(72) Inventor: **NIYOGI, Suhas, G.**; 24 Clark Ridge,  
Hockessin, DE 19707 (US).

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU,  
AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU,

CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM,  
HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK,  
LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX,  
MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL,  
TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW.

(84) Designated States (*regional*): ARIPO patent (GH, GM,  
KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian  
patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European  
patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE,  
IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF,  
CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

**Declaration under Rule 4.17:**

— as to the applicant's entitlement to claim the priority of the  
earlier application (Rule 4.17(iii)) for all designations

**Published:**

— without international search report and to be republished  
upon receipt of that report

For two-letter codes and other abbreviations, refer to the "Guid-  
ance Notes on Codes and Abbreviations" appearing at the begin-  
ning of each regular issue of the PCT Gazette.

WO 01/96467 A2

(54) Title: COMPOSITE MATERIALS COMPRISING PROPYLENE GRAFT COPOLYMERS

(57) Abstract: A composite material comprises a smectite clay having exchangeable cations that has been treated with at least one organic swelling agent, uniformly dispersed in a graft copolymer having a backbone of a porous propylene polymer material, to which is graft polymerized at least one grafting monomer capable of being polymerized by free radicals, wherein the total inorganic content of the composite material is 0.5 % to 10 %, based on the total weight of the composite material. The composite material is made by graft polymerizing at least one liquid monomer capable of being polymerized by free radicals onto a porous propylene polymer material, in a non-oxidizing environment, in the presence of a smectite clay and an organic free radical polymerization initiator, whereby the chains of polymerized monomer that are formed intercalate the clay and produce a uniform dispersion of clay particles within the particulate propylene polymer material.

## COMPOSITE MATERIALS COMPRISING PROPYLENE GRAFT COPOLYMERS

### Field of the Invention

This invention relates to composite materials comprising a graft copolymer of a propylene polymer material, and a smectite-type clay that has been treated with a swelling agent.

### Background of the Invention

Layered clay minerals such as montmorillonite are composed of silicate layers with a thickness on a nanometer scale (1 nanometer = 10Å). Dispersion of such layered materials in polymers are frequently referred to as nanocomposites.

It is known that these silicates, such as smectite clays (e.g., sodium and calcium montmorillonite), can be treated with organic swelling agents such as organic ammonium ions, to intercalate the swelling agent molecules between adjacent, planar silicate layers, thereby substantially increasing the interlayer spacing and making the clay more hydrophobic and better suited for interaction with polymer precursors. The intercalated silicates can then be exfoliated, i.e., the silicate layers are separated, typically by high shear mixing. The individual silicate layers, when admixed with a matrix polymer before, after, or during the polymerization of the matrix polymer, have been found to substantially improve one or more properties of the polymer such as mechanical strength and/or high temperature characteristics.

For example, US 4,810,734 discloses a process for producing a composite material by contacting a layered clay mineral with a swelling agent in the presence of a dispersion medium such as water, an alkanol, or dimethyl sulfoxide, mixing with a molten polymerizable monomer or a mixture of monomer and dispersion medium, and polymerizing the monomer in the mixture. Catalysts and accelerators for polymerization can also be present. The polymer that is formed can be, for example, a polyamide, a vinyl polymer, or a thermoset resin.

US 5,514,734 discloses a composite material comprising a polymer matrix having layered or fibrillar particles, e.g., phyllosilicates, uniformly dispersed therein, the particles being bonded to organosilanes, organo titanates, or organo zirconates and

having one or more moieties bonded to at least one polymer in the polymer matrix. The polymer matrix can be, for example, a polyester, polyolefin, or polyamide.

US 5,760,121 discloses a composite material comprising a host material such as a polyamide, polyvinylamine, polyethylene terephthalate, polyolefin, or polyacrylate, and  
5 exfoliated platelets of a phyllosilicate material. The platelets are derived from an intercalate formed without an onium ion or silane coupling agent by contacting with an intercalant polymer-containing composition containing water and/or an organic solvent. Suitable intercalant polymers include polyolefins and acrylic polymers.

US 5,910,523 discloses a composition comprising (a) a semi-crystalline  
10 polyolefin, (b) a clay filler having dispersible platelets in stacks, (c) an amino-functional silane reacted with the filler, and (d) a carboxylated or maleated semi-crystalline polyolefin that has been reacted with the aminofunctional silane after the silane was reacted with the filler.

Incorporating clay minerals into a polymer matrix, however, does not always  
15 result in markedly improved mechanical properties. This may be due in part to the inability to exfoliate all or at least a substantial portion of the layers of the silicate material. It may also be due in part to the lack of affinity between the layered silicate materials and the organic polymers. Attempts to overcome these problems have not been totally successful. There is therefore a need for novel composite materials that have  
20 desirable mechanical and physical properties.

#### Summary of the Invention

The composite material of this invention comprises:

- (1) a graft copolymer comprising a backbone of a particulate propylene polymer material having a weight average diameter of about 0.4-7 mm, a surface area of at least 0.1  
25 m<sup>2</sup>/g, and a pore volume fraction of at least about 0.07 and wherein more than 40% of the pores in the particle have a diameter greater than 1 micron, to which is graft polymerized at least one grafting monomer capable of being polymerized by free radicals, and
- (2) a smectite clay having exchangeable cations that has been treated with at least one  
30 organic swelling agent, uniformly dispersed in the particulate propylene polymer material, wherein the total inorganic content of the composite material is about 0.5% to about 10%, based on the total weight of the composite.

The composite material of this invention is made by a process comprising, in a substantially non-oxidizing environment, graft polymerizing at least one liquid monomer capable of being polymerized by free radicals to a particulate propylene polymer material having a weight average diameter of about 0.4-7 mm, a surface area of at least 0.1 m<sup>2</sup>/g, and a pore volume fraction of at least about 0.07 and wherein more than 40% of the pores in the particle have a diameter greater than 1 micron, in the presence of (1) a smectite clay having exchangeable cations that has been treated with at least one organic swelling agent, and (2) about 0.1 parts to about 6 parts per hundred parts of the propylene polymer material of an organic free radical polymerization initiator, at a reaction temperature of about 60°C to about 125°C, whereby the chains of polymerized monomer that are formed intercalate the clay and produce a uniform dispersion of clay particles within the particulate propylene polymer material, the composite material having an inorganic content of about 0.5% to about 10%, based on the total weight of the composite.

The polymer composite materials produced according to the method of this invention have excellent mechanical properties such as heat distortion temperature, tensile strength, and flexural modulus.

#### Detailed Description of the Invention

The propylene polymer material that is used as the backbone of the graft copolymer in the composite material of this invention can be:

- (1) a homopolymer of propylene having an isotactic index greater than 80, preferably about 85 to about 99;
- (2) a copolymer of propylene and an olefin selected from the group consisting of ethylene and C<sub>4-10</sub> alpha-olefins, provided that when the olefin is ethylene, the maximum polymerized ethylene content is about 10%, preferably about 4%, and when the olefin is a C<sub>4-10</sub> alpha-olefin, the maximum polymerized content thereof is about 20% by weight, preferably about 16%, the copolymer having an isotactic index greater than 85;
- (3) a terpolymer of propylene and two olefins selected from the group consisting of ethylene and C<sub>4-8</sub> alpha-olefins, provided that the maximum polymerized C<sub>4-8</sub> alpha-olefin content is 20% by weight, preferably about 16%, and, when ethylene is one of the olefins, the maximum polymerized ethylene content is 5% by

weight, preferably about 4%, the terpolymer having an isotactic index greater than 85;

(4) an olefin polymer composition comprising:

(a) about 10% to about 60% by weight, preferably about 15% to about 55%, of a propylene homopolymer having an isotactic index greater than 80, preferably about 85 to about 98, or a copolymer of monomers selected from the group consisting of (i) propylene and ethylene, (ii) propylene, ethylene and a C<sub>4-8</sub> alpha-olefin, and (iii) propylene and a C<sub>4-8</sub> alpha-olefin, the copolymer having a polymerized propylene content of more than 85% by weight, preferably about 90% to about 99%, and an isotactic index greater than 85;

(b) about 5% to about 25% by weight, preferably about 5% to about 20%, of a copolymer of ethylene and propylene or a C<sub>4-8</sub> alpha-olefin that is insoluble in xylene at room temperature; and

(c) about 30% to about 70% by weight, preferably about 40% to about 65%, of an elastomeric copolymer of monomers selected from the group consisting of (i) ethylene and propylene, (ii) ethylene, propylene, and a C<sub>4-8</sub> alpha-olefin, and (iii) ethylene and a C<sub>4-8</sub> alpha-olefin, the copolymer optionally containing about 0.5% to about 10% by weight of a polymerized diene and containing less than 70% by weight, preferably about 10% to about 60%, most preferably about 12% to about 55%, of polymerized ethylene and being soluble in xylene at room temperature and having an intrinsic viscosity, measured in decahydronaphthalene at 135°C, of about 1.5 to about 4.0 dl/g,

wherein the total amount of (b) and (c), based on the total olefin polymer composition, is about 50% to about 90%, the weight ratio of (b)/(c) is less than 0.4, preferably 0.1 to 0.3, and the composition is prepared by polymerization in at least two stages and has a flexural modulus of less than 150 MPa; or

(5) a thermoplastic olefin comprising:

(a) about 10% to about 60%, preferably about 20% to about 50%, of a propylene homopolymer having an isotactic index greater than 80, or a copolymer of monomers selected from the group consisting of (i) ethylene

and propylene, (ii) ethylene, propylene and a C<sub>4-8</sub> alpha-olefin, and (iii) ethylene and a C<sub>4-8</sub> alpha-olefin, the copolymer having a polymerized propylene content greater than 85% and an isotactic index of greater than 85;

5 (b) about 20% to about 60%, preferably about 30% to about 50%, of an amorphous copolymer of monomers selected from the group consisting of (i) ethylene and propylene, (ii) ethylene, propylene, and a C<sub>4-8</sub> alpha-olefin, and (iii) ethylene and a C<sub>4-8</sub> alpha-olefin, the copolymer optionally containing about 0.5% to about 10% of a polymerized diene, and  
10 containing less than 70% polymerized ethylene and being soluble in xylene at room temperature; and

(c) about 3% to about 40%, preferably about 10% to about 20%, of a copolymer of ethylene and propylene or a C<sub>4-8</sub> alpha-olefin that is insoluble in xylene at room temperature,

15 wherein the thermoplastic olefin has a flexural modulus of greater than 150 but less than 1200 MPa, preferably about 200 to about 1100 MPa, and most preferably about 200 to about 1000 MPa.

Room temperature is about 25°C.

The C<sub>4-8</sub> alpha-olefins useful in the preparation of (4) and (5) include, for  
20 example, butene-1, pentene-1; hexene-1; 4-methyl-1-pentene, and octene-1.

The diene, when present, is typically a butadiene; 1,4-hexadiene; 1,5-hexadiene, or ethylidenenorbornene.

Propylene polymer materials (4) and (5) can be prepared by polymerization in at least two stages, where in the first stage the propylene; propylene and ethylene;  
25 propylene and an alpha-olefin, or propylene, ethylene and an alpha-olefin are polymerized to form component (a) of (4) or (5), and in the following stages the mixtures of ethylene and propylene; ethylene and the alpha-olefin, or ethylene, propylene and the alpha-olefin, and optionally a diene, are polymerized to form components (b) and (c) of (4) or (5).

30 The polymerization can be conducted in liquid phase, gas phase, or liquid-gas phase using separate reactors, all of which can be done either by batch or continuously. For example, it is possible to carry out the polymerization of component (a) using liquid

propylene as a diluent, and the polymerization of components (b) and (c) in gas phase, without intermediate stages except for the partial degassing of the propylene. All gas phase is the preferred method.

The preparation of propylene polymer material (4) is described in more detail in  
5 US Patents 5,212,246 and 5,409,992. The preparation of propylene polymer material (5) is described in more detail in US Patents 5,302,454 and 5,409,992.

Propylene homopolymer is the preferred propylene polymer backbone material.

The propylene polymer material used as the backbone of the graft copolymer is in particulate form and has a weight average diameter of about 0.4-7 mm, a surface area of  
10 at least  $0.1 \text{ m}^2/\text{g}$ , and a pore volume fraction of at least about 0.07, i.e., seven percent of the volume is pores, and wherein more than 40% of the pores in the particle have a diameter greater than 1 micron. Preferably more than 50% of the pores in the particle have a diameter greater than 1 micron, and most preferably more than 90% of the pores in the particle have a diameter greater than 1 micron. The pore volume fraction is  
15 preferably at least 0.12, most preferably at least 0.20.

The grafting monomers useful in accordance with this invention can be any monomeric vinyl compounds capable of being polymerized by free radicals, where the vinyl radical  $\text{H}_2\text{C}=\text{CR}-$ , in which  $\text{R} = \text{H}$  or methyl, is attached to a straight or branched aliphatic chain or to a substituted or unsubstituted aromatic, heterocyclic, or alicyclic ring  
20 in a mono- or polycyclic compound. Typical substituent groups can be alkyl, hydroxyalkyl, aryl, and halo. Usually the vinyl monomer will be a member of one of the following classes: (1) vinyl-substituted aromatic, heterocyclic, or alicyclic compounds, including styrene, vinylnaphthalene, vinylpyridine, vinylpyrrolidone, vinylcarbazole, and homologs thereof, e.g., alpha- and para-methylstyrene, methylchlorostyrene, p-tert-  
25 butylstyrene, methylvinylpyridine, and ethylvinylpyridine; (2) vinyl esters of aromatic and saturated aliphatic carboxylic acids, including vinyl formate, vinyl acetate, vinyl chloroacetate, vinyl cyanoacetate, vinyl propionate, and vinyl benzoate; and (3) unsaturated aliphatic nitriles and carboxylic acids and their derivatives, including acrylonitrile; methacrylonitrile; acrylamide; methacrylamide, and acrylic acid; acrylate  
30 esters, such as the methyl, ethyl, hydroxyethyl, 2-ethylhexyl, and butyl acrylate esters; methacrylic acid; ethacrylic acid, and methacrylate esters such as the methyl, ethyl, butyl,

benzyl, phenylethyl, phenoxyethyl, epoxypentyl, and hydroxypentyl methacrylate esters, and maleic anhydride. One or more of these monomers can be used.

The total amount of polymerized monomers is about 10 to about 100 parts, preferably about 25 to about 95 parts, per hundred parts of the propylene polymer material.

The smectite clays that are the second component of the composite materials of this invention are layered silicate minerals having exchangeable cations that can be treated with organic swelling agents such as organic ammonium ions, to intercalate the organic molecules between adjacent planar silicate layers, thereby substantially increasing the interlayer spacing. The expansion of the interlayer distance of the layered silicate facilitates the intercalation of the clay with other materials, in this case the polymerizable grafting monomer. The interlayer spacing of the silicates can be further increased by formation of the polymerized monomer chains between the silicate layers. The intercalated silicate platelets act as a nanoscale (sub-micron size) filler for the polymer.

Intercalation of the silicate layers in the clay can take place either by cation exchange or by absorption. For intercalation by absorption, dipolar functional organic molecules such as nitrile, carboxylic acid, hydroxy, and pyrrolidone groups must be present on the clay surface. Intercalation by absorption can take place when either acid or non-acid clays are used as the starting material. Cation exchange takes place only if an ionic clay containing ions such as, for example,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Ca}^{++}$ ,  $\text{Ba}^{++}$ , and  $\text{Li}^+$  is used. Acid clays are not suitable starting materials if intercalation by cation exchange is desired. Ionic clays can also absorb dipolar organic molecules.

Smectite clays include, for example, montmorillonite, saponite, beidellite, hectorite, and stevensite, where the space between silicate layers is typically 17-36 Å, measured by small angle X-ray scattering. Typically, a clay with exchangeable cations such as sodium, calcium and lithium ions is used. Montmorillonite in the sodium exchanged form is preferred.

The organic swelling agent used to treat the clay is a quaternary ammonium compound, excluding pyridinium ion, such as, for example, poly(propylene glycol)bis(2-aminopropyl ether), poly(vinylpyrrolidone), dodecylamine hydrochloride,



octadecylamine hydrochloride, and dodecylpyrrolidone. Such treated clays are commercially available. One or more of these swelling agents can be used.

In the process of this invention the organoclay, the liquid monomer, and the polymerization initiator pass into the pores of the particulate propylene polymer material and the intercalation of the liquid monomer between the clay layers and subsequent graft polymerization reaction take place within the particles of propylene polymer material. While not intending to be bound by theory, it is believed that exfoliation (delamination), i.e., the separation of the silicate layers into individual silicate platelets, can take place when the particles of graft copolymer containing the intercalated clay particles are extruded to form useful articles of manufacture.

The total amount of inorganic material from the clay mineral in the composite product is about 0.5% to about 15%, preferably about 0.5% to about 10%, based on the total weight of the composite. The amount of clay used is limited by the extremely high viscosity of the clay/monomer suspension.

Optionally, the composite material can be blended with about 2% to about 45%, preferably about 5% to about 30%, based on the total weight of the composition, of one or more rubber components, and/or about 5% to about 90%, preferably about 10% to about 70%, based on the total weight of the composition, of a broad molecular weight distribution propylene polymer material.

The rubber component is selected from one or more of the group consisting of (i) an olefin copolymer rubber, (ii) a monoalkenyl aromatic hydrocarbon-conjugated diene block copolymer, and (iii) a core-shell rubber. Any of these rubber components can have acid or anhydride functionality or can be free of these functional groups. The preferred rubber components are (i) or (ii), either alone or in combination.

Suitable olefin copolymer rubbers include, for example, saturated olefin copolymer rubbers such as ethylene/propylene monomer rubbers (EPM), ethylene/octene-1, and ethylene/butene-1 rubbers, and unsaturated olefin copolymer rubbers such as ethylene/propylene/diene monomer rubbers (EPDM). The preferred olefin copolymer rubbers are ethylene/propylene, ethylene/butene-1, and ethylene/octene-1 copolymer rubbers.

The monoalkenyl aromatic hydrocarbon-conjugated diene block copolymer can be a thermoplastic elastomer of the A-B (or diblock) structure, the linear A-B-A (or

triblock) structure, the radial (A-B)<sub>n</sub> type where n = 3-20%, or a combination of these structure types, wherein each A block is a monoalkenyl aromatic hydrocarbon polymer block, and each B block is an unsaturated rubber block. Various grades of copolymers of this type are commercially available. The grades differ in structure, molecular weight of the mid and end blocks, and the ratio of monoalkenyl aromatic hydrocarbon to rubber. The block copolymer can also be hydrogenated. Typical monoalkenyl aromatic hydrocarbon monomers are styrene, ring-substituted C<sub>1-4</sub> linear or branched alkyl styrenes, and vinyltoluene. Styrene is preferred. Suitable conjugated dienes include, for example, butadiene and isoprene. Preferred block copolymers are hydrogenated styrene/ethylene-butene-1/styrene triblock copolymers.

The weight average molecular weight (M<sub>w</sub>) of the block copolymers generally will be in the range of about 45,000 to about 260,000 g/mole, average molecular weights in the range of about 50,000 to about 125,000 g/mole being preferred because they produce compositions having the best balance of impact strength and stiffness. Also, while block copolymers having unsaturated as well as saturated rubber blocks can be used, copolymers having saturated rubber blocks are preferred, also on the basis of the impact/stiffness balance of the compositions containing them. The weight ratio of monoalkenyl aromatic hydrocarbon to conjugated diene rubber in the block copolymer is in the range of about 5/95 to about 50/50, preferably about 10/90 to about 40/60.

The core-shell rubber components comprise small particles of a crosslinked rubber phase surrounded by a compatibilizing shell, normally a glassy polymer or copolymer. The core is typically a diene rubber such as butadiene or isoprene rubber, or a polyacrylate. The shell is typically a polymer of two or more monomers selected from styrene, methyl methacrylate, and acrylonitrile. Particularly preferred core-shell rubbers have a polyacrylate core.

Suitable rubbers useful as impact modifiers include, for example, Engage 8100 and 8150 ethylene/octene-1 copolymers, commercially available from DuPont-Dow Elastomers; EPM 306P ethylene/propylene copolymer, commercially available from the Polysar Rubber Division of Miles, Incorporated; and Kraton RP6912 styrene/ethylene-propylene/styrene triblock copolymer rubber and Kraton FG1901X styrene/ethylene-butene-1/styrene triblock copolymer rubber modified with maleic anhydride, commercially available from Shell Chemical Company. A thermoplastic olefin such as

the one described under (5) of the propylene polymer materials suitable for use as the backbone polymer of the graft copolymer can also be used as an impact modifier.

Another optional component that can be blended with the composite material of this invention is a broad molecular weight distribution propylene polymer material (BMWD PP) having a  $M_w/M_n$  of about 5 to about 60, preferably about 5 to about 40; a melt flow rate of about 0.5 to about 50 g/10 min, preferably about 1 to about 30 g/10 min, and xylene insolubles at 25°C of greater than or equal to 94%, preferably greater than or equal to 96%, and most preferably greater than or equal to 98%. The propylene polymer material having a broad molecular weight distribution can be a homopolymer of propylene or an ethylene/propylene rubber impact-modified homopolymer of propylene, wherein the propylene homopolymer has a broad molecular weight distribution.

The BMWD PP can be prepared by sequential polymerization in at least two stages, in the presence of a Ziegler-Natta catalyst supported on magnesium halide in active form. The polymerization process occurs in separate and consecutive stages, and in each stage polymerization takes place in the presence of the polymer and the catalyst from the preceding stage.

The polymerization process can be carried out in a batch or in a continuous mode according to known techniques, operating in liquid phase in the presence or not of an inert diluent, or in gas phase, or liquid-gas phase, preferably in gas phase. The preparation of the BMWD PP is described in more detail in US Patent 5,286,791.

Other additives such as pigments, slip agents, waxes, oils, antiblocking agents and antioxidants can also be present in the composition.

The composite material of this invention is made by a process comprising, in a substantially non-oxidizing environment, graft polymerizing at least one liquid monomer capable of being polymerized by free radicals to a particulate propylene polymer material having a weight average diameter of about 0.4-7 mm, a surface area of at least 0.1 m<sup>2</sup>/g, and a pore volume fraction of at least 0.07, preferably at least 0.12, and most preferably at least 0.20, and wherein more than 40% of the pores in the particle have a diameter greater than 1 micron, preferably more than 50% and most preferably more than 90%, in the presence of (1) a smectite clay having exchangeable cations that has been treated with at least one organic swelling agent, and (2) about 0.1 to about 6 parts, preferably about 0.2 to about 3 parts, per hundred parts of the propylene polymer material of an organic

free radical polymerization initiator at a reaction temperature of about 60°C to about 125°C, whereby the chains of polymerized monomer that are formed intercalate the clay particles and produce a uniform dispersion of clay particles within the particulate propylene polymer material, the composite material having an inorganic content of about 5 0.5% to about 15%, preferably about 0.5% to about 10%, based on the total weight of the composition.

“Non-oxidizing environment” is the environment or atmosphere to which the olefin polymer material is exposed during the preparation of the graft copolymer and means an environment in which the active oxygen concentration, i.e., the concentration 10 of oxygen in a form that will react with the free radicals in the polymer material, is less than 15%, preferably less than 5%, and most preferably less than 1% by volume. The most preferred concentration of active oxygen is 0.004% or lower by volume. Within these limits, the non-oxidizing atmosphere can be any gas, or mixture of gases, that is oxidatively inert toward the free radicals in the olefin polymer material, e.g., nitrogen, 15 argon, helium, and carbon dioxide.

Free radical or active sites are produced in the particulate propylene polymer material by treating the propylene polymer material with an organic compound that is a free-radical-generating polymerization initiator. The polymerization initiator is one that has a decomposition half-life at the temperature employed of about 1 to about 240, 20 preferably about 5 to about 100, and most preferably about 10 to about 40, minutes. Organic peroxides, and especially those that generate alkoxy radicals, constitute the preferred class of initiators. These include acyl peroxides, such as benzoyl and dibenzoyl peroxides; dialkyl and aralkyl peroxides, such as di-tert-butyl peroxide, dicumyl peroxide, cumyl butyl peroxide, 1,1-di-tert-butylperoxy-3,5,5-trimethylcyclohexane, 2,5- 25 dimethyl-2,5-di-tert-butylperoxyhexane, and bis(alpha-tert-butyl peroxyisopropyl-benzene); peroxy esters, such as tert-butylperoxy pivalate, tert-butyl perbenzoate, tert-butyl peroctoate; 2,5-dimethylhexyl 2,5-di(perbenzoate), tert-butyl di(perphthalate), tert-butylperoxy-2-ethyl hexanoate; and 1,1-dimethyl-3-hydroxybutylperoxy-2-ethyl hexanoate; and peroxy carbonates, such as di(2-ethylhexyl) peroxy dicarbonate, di(n- 30 propyl)peroxy dicarbonate, and di(4-tert-butylcyclohexyl)peroxy dicarbonate. Peroxy esters are preferred.

Preparation of graft copolymers by contacting a propylene polymer material with a free radical polymerization initiator such as an organic peroxide and a vinyl monomer is described in more detail in US 5,140,074.

5 In one embodiment of the process of this invention, the particulate propylene polymer material is heated to the reaction temperature, the organic free radical initiator is added, and a suspension of the clay in liquid monomer is then added separately.

In a second embodiment of the process of this invention, the particulate propylene polymer material is mixed with a suspension of the clay in liquid monomer, the mixture is heated to the reaction temperature, and the organic free radical initiator is added.

10 In a third embodiment, a suspension of the clay in liquid monomer and the organic free radical initiator are added concurrently to the particulate propylene polymer material, which has been heated to the reaction temperature.

In another embodiment, the clay is mixed with the particulate propylene polymer material in the presence of an hydrocarbon solvent (preferably aliphatic), the mixture is  
15 heated to the reaction temperature to remove the solvent, and the organic free radical initiator and liquid monomer are added concurrently. Suitable hydrocarbon solvents include pentane, heptane, and C<sub>8-12</sub> linear or branched aliphatic hydrocarbons.

In preparing the organoclay/monomer suspensions in the process of this invention, the clay and the organic swelling agent are mixed, which produces a  
20 hydrophobic organoclay. The hydrophobic clay is then dried and pulverized. Monomer is added to the pulverized organoclay with stirring for a prolonged period of time. Suspending or thickening agents such as polyvinyl alcohol can also be added during the stirring process.

We have found that simply mixing a preformed graft copolymer with an  
25 organoclay, rather than graft polymerizing in the presence of an organoclay, does not produce a significant improvement in the mechanical properties of the product.

The composite materials made by the process of this invention have improved tensile strength, flexural modulus, and heat stability. They can be used to make articles of manufacture by conventional shaping processes such as melt spinning, casting,  
30 vacuum molding, sheet molding, injection molding and extruding. Examples of such articles are components for technical equipment, household equipment, sports equipment, bottles, containers, components for the electrical and electronics industries, automobile

components, and fibers. They are especially useful for the fabrication of extruded films and film laminates, for example, films for use in food packaging.

The test methods used to evaluate the composites of this invention include:

	Flexural modulus	ASTM D-790-86
5	Tensile strength	ASTM D-638-89
	Elongation at yield	ASTM D-638-89
	Melt flow rate (propylene polymer materials) (230°C, 2.16 kg)	ASTM D-1238
	Dual gated tensile strength	ASTM D-638-89
10	Flexural strength	ASTM D-618, Procedure A
	Notched Izod impact	ASTM D-256-87
	Rockwell hardness, R	ASTM D-785
	Heat distortion temperature	ASTM D-648
	Strength @ break	ASTM D-638-89
15	2% Secant modulus	ASTM D-882

Isotactic index is defined as the per cent of olefin polymer insoluble in xylene.

The weight percent of olefin polymer soluble in xylene at room temperature is determined by dissolving 2.5 g of the polymer in 250 ml of xylene in a vessel equipped with a stirrer, that is heated at 135°C with agitation for 20 minutes. The solution is cooled to 25°C while continuing the agitation, and then left to stand without agitation for 30 minutes so that the solids can settle. The solids are filtered with filter paper, the remaining solution is evaporated by treating it with a nitrogen stream, and the solid residue is vacuum dried at 80°C until a constant weight is reached. The percent by weight of polymer insoluble in xylene at room temperature is the isotactic index of the polymer. The value obtained in this manner corresponds substantially to the isotactic index determined via extraction with boiling n-heptane, which by definition constitutes the isotactic index of the polymer.

Intrinsic viscosity is measured in decahydronaphthalene at 135°C.

The pore volume fraction values were determined by a mercury porosimetry technique in which the volume of mercury absorbed by the particles is measured. The volume of mercury absorbed corresponds to the volume of the pores. This method is described in Winslow, N. M. and Shapiro, J. J., "An Instrument for the Measurement of

Pore-Size Distribution by Mercury Penetration," *ASTM Bull.*, TP 49, 39-44 (Feb. 1959), and Rootare, H. M., "A Review of Mercury Porosimetry," 225-252 (in Hirshhom, J. S. and Roll, K. H., Eds., *Advanced Experimental Techniques in Powder Metallurgy*, Plenum Press, New York, 1970).

5           The surface area measurements were made by the B.E.T. method as described in *JACS* 60, 309 (1938).

In this specification, all parts and percentages are by weight unless otherwise noted.

#### Example 1

10           This example describes the preparation and properties of composites comprising polypropylene grafted with polystyrene in the presence of clays that have been treated with various swelling agents.

The clay used in the preparation of organoclays A and B was Montmorillonite K10, an acid clay that is commercially available from Aldrich Chemical Company. The  
15    clay had the following properties: surface area = 220-270 m<sup>2</sup>/g; bulk density = 300-370 g/l, and pore volume = 0.29 cm<sup>3</sup>/g.

Organoclay A was prepared by suspending 30 g of montmorillonite K10 in 200 ml of deionized water and heating to 60°C. In a separate beaker, 15 g of poly(propylene glycol)bis(2-aminopropyl ether) were dissolved in 100 ml of water and heated to 70-  
20    75°C. 37% HCl (12 g) was added slowly while stirring. After two hours, the solution was poured into the clay suspension maintained at 60°C and stirred for two hours at that temperature. The resulting organoclay was filtered, washed neutral, air dried, and finally dried at 60°C under vacuum. The final weight was 38 g. Intercalation of the silicate layers of the clay with the organic swelling agent took place by absorption.

25           Organoclay B was prepared by suspending 110 g of montmorillonite K10 in 500 ml of water, and heating to 50°C with a 3% aqueous solution of NaHCO<sub>3</sub> for four hours. In a separate beaker, 40 g of dodecylamine and 2 g of poly(propylene glycol)bis(2-aminopropyl ether) were suspended in 200 ml of water at 80°C. A solution of 30% HCl (20 ml) was added slowly to obtain a clear solution. This clear solution was added to the  
30    clay suspension at 60°C and stirred for two hours. The clay was filtered, washed neutral, and dried to obtain 123 g of product. Intercalation of the silicate layers of the clay with the organic swelling agents took place by ion exchange.

Composites A and B were prepared using a 2 liter glass jacketed reactor fitted with an air driven helical stirrer, a thermocouple, nitrogen inlet and outlet, and monomer and peroxide addition needles.

5 The propylene homopolymer used as the backbone polymer in the preparation of the graft copolymer had the following properties: a melt flow rate of 9 kg/min, a pore volume fraction of 0.28, a surface area of 0.3 m<sup>2</sup>/g, and greater than 40% of the pores with a diameter greater than one micron, commercially available from Montell USA Inc.

Composite A was prepared by suspending organoclay A (30 g) in styrene (140 g) and purging with nitrogen at 10°C for 15 minutes. The propylene homopolymer (350 g)  
10 was charged to the reactor and purged with nitrogen at room temperature for two hours and at 40°C for two hours. The reactor temperature was reduced to about 30°C. A deaerated solution of 4.4 g of tert-butylperoxy pivalate in 80 ml of cold pentane was fed to the reactor over a period of ten minutes and pentane was driven out with a flow of nitrogen and was collected in a cold trap. The reactor was heated to 64°C for 35-40  
15 minutes under a slow flow of nitrogen. The nitrogen flow was cut off and the clay suspension in styrene was added slowly over a period of thirty minutes. The styrene add level was 40 parts styrene per hundred parts of propylene homopolymer. The reactor temperature increased to 78°C from the heat of polymerization and was maintained at 75°-78°C for five hours. The reaction mixture was cooled under nitrogen, then polymer  
20 was taken out and subjected to vacuum at 60°C. The final weight of the product was 510g.

Composite B was prepared in the manner described above except that organoclay B was used in place of organoclay A.

The graft copolymer/clay composites were dry mixed in a plastic bag with the  
25 amounts of rubber and stabilizer shown in Table 1 prior to extrusion. The mixture was extruded on a Haake twin screw extruder at 232°C and molded on a Boy molding machine at 215°C to prepare the samples for physical testing.

The rubber was Engage 8100 ethylene/octene-1 rubber containing 24% octene-1, commercially available from DuPont-Dow Elastomers. The stabilizer was a 2.5/1 blend  
30 of Irganox 215 antioxidant and calcium stearate. Irganox 215 antioxidant is a mixture of 1 part Irganox 1010 antioxidant and 2 parts Irgafos 168 tris(2,4-di-tert-butylphenyl) phosphite stabilizer, commercially available from Ciba Specialty Chemicals Corporation.



The propylene homopolymer grafted with polystyrene in the Control was prepared as described under Composite A, except that no clay was present. An attempt to suspend the clay in styrene monomer for grafting was unsuccessful. The graft copolymer, montmorillonite K10 (designated as Mont. K10 in Table 1), rubber, and stabilizer in the amounts shown in Table 1 were dry mixed prior to extrusion. The clay was not treated with a swelling agent. The dry mixed materials were extruded on a Haake twin screw extruder at 230°C (450°F) at 100 rpm.

The composition of the samples and the results of the physical property testing are given in Table 1. In Table 1, % polystyrene (PSt), % propylene homopolymer (PP), % rubber, % organoclay, and % inorganic content all refer to the % by weight of the final product.

<b>Table 1</b>			
<b>Sample</b>	<b>1</b>	<b>2</b>	<b>Control</b>
Composite A (g)	480	-	-
Composite B (g)	-	480	-
Engage 8100 rubber (g)	35	35	35
Stabilizer (g)	0.9	0.9	0.9
PP-g-styrene (g)	-	-	450
Mont. K10 (g)	-	-	30
PSt (%)	25.1	25.19	28.58
PP (%)	62.72	62.74	71.42
Engage 8100 rubber (%)	6.80	6.80	6.80
Organoclay (%)	5.38	5.38	5.82
Inorganic content (%)	3.88	3.89	5.82
Tensile strength (MPa)	31.2	32.3	31.3
Elong. @ yield (%)	3.1	3.3	3.6
Strength @ break (MPa)	27.4	26.2	25.6
Flex. strength (MPa)	49.8	51.5	49.3
Flex. modulus (MPa)	1807	1908	1732
Izod impact @ 23°C (ft.lb/in)	0.43	0.49	0.60
Hardness – R	97	-	93
HDT @ 455 kPa (°C)	100	-	96

The data show that the composite materials of this invention have a better balance of physical properties than the control, in which the preformed graft copolymer was mixed with a clay that was not treated with an organic swelling agent.

#### Example 2

- 5 This example describes the preparation and properties of composites comprising polypropylene grafted with polystyrene in the presence of a commercial organoclay.

The clay/monomer suspension in Sample 1 was EA 2782, a suspension in styrene monomer of montmorillonite that was treated with a swelling agent, supplied by RHEOX, Inc.

5 The composite described in Sample 1 was prepared as follows. The propylene homopolymer described in Example 1 (400 g) was purged with nitrogen and heated to 115°C under a nitrogen purge. The suspension of clay in styrene monomer (195 g) was fed to the reactor described in Example 1 at a rate of about 4.1 g/min. In Table 2, pph refers to the parts of styrene added per hundred parts of propylene homopolymer. Tert-butyl peroctaoate (8.32 g) was diluted with 25 ml of n-octane and fed into the reactor  
10 over a period of 15 minutes. The polymerization was continued for another two hours at a reactor pressure of 40 kPa (6 psi). The temperature was raised to 125°C and purged with nitrogen for 60 minutes. After cooling, 590 g of product was removed from the reactor.

15 The graft copolymer used in Control Example 1 was prepared as described above, except that clay was not present.

The graft copolymer in Sample 1 and in the control was stabilized with the same stabilizer as in Example 1. The stabilized graft copolymer/clay composite and the control without clay were then extruded on a Haake twin screw extruder at 230°C (450°F) and molded on a Boy molding machine at a temperature of 230°C (450°F).

20 The composition of the samples and the results of the physical property testing are given in Table 2.

Table 2		
Sample 1	1	Control 1
Styrene (pph)	20.0	21.5
Clay (% in monomer)	8	-
Stabilizer (%)	0.2	0.2
Inorganic content of product (%)	1.7	0.0
Notched Izod impact (ft-lb/in)	0.44	0.36
Tensile strength (kpsi)	5.80	5.54
Elong. @ yield (%)	4.3	4.9
2% Secant modulus (kpsi)	298	262

The data show that although only a small amount of clay was present in the product (1.7%), the physical properties of the composite material were better than those of the control without clay.

### Example 3

5 This example describes a larger scale preparation and the properties of composites comprising polypropylene grafted with polystyrene (PP-g-St) or poly(methyl methacrylate) (PP-g-PMMA) in the presence of a commercial organoclay.

The organoclays were supplied as a suspension in styrene or methyl methacrylate of montmorillonite that had been treated with a swelling agent, available from RHEOX,  
10 Inc.

The composite described in Sample 1 of Table 3 was prepared as follows. The propylene homopolymer (1.3 kg; 3.5 lb) described in Example 1 was charged to a reactor and the reactor was purged with nitrogen for about one hour at room temperature. The reactor was heated to 120°C under a nitrogen purge, after which the purge was

terminated. The organoclay/styrene suspension EA-2780 (600 g) was diluted with 200 g of styrene monomer under a nitrogen purge with stirring. Tert-butyl peroctoate (28 g of 50% solution in odorless mineral spirits diluted with 50 ml of heptane) and the clay/styrene suspension were concurrently fed to the reactor at a rate of 2.5 ml/min and 15 ml/min, respectively. The total addition time was 90 minutes. The reaction was continued for one more hour and the reactor was purged with nitrogen to cool. The 1.88 kg (5.04 lb) of product contained 45 parts of styrene per hundred parts of propylene homopolymer.

The polypropylene grafted with polystyrene in Control 1 was prepared by grafting polystyrene onto the propylene homopolymer described in Example 1 at a grafting temperature of 120°C. Forty-five parts by weight of styrene were added per 100 parts of propylene homopolymer (pph). Lupersol PMS 50% t-butylperoxy-2-ethyl hexanoate in mineral spirits, commercially available from Elf Atochem, was used as the peroxide initiator. The monomer was fed at 0.9 pph/min, and a monomer to initiator molar ratio of 105 was used. After completion of monomer and peroxide addition, the temperature was raised to 140°C for 60 minutes under a nitrogen purge.

The composite described in Sample 2 was prepared by mixing in the reactor 1.27 kg (3.4 lb) of the propylene homopolymer described in Example 1 and 1.31 kg (3.5 lb) of a suspension of the organoclay containing 9.53% methyl methacrylate under a nitrogen atmosphere at room temperature for 45 minutes. The mixture was heated to 95°C and a solution of 28 g tert-butyl peroxy-pivalate (75% solution in odorless mineral spirits) diluted with 150 ml odorless mineral spirits was fed to the reactor at a rate of 6 ml/min. The polymerization was carried out at 95°C for one hour after all of the solution was added. The reactor was heated to 100°C and the pressure was released. The reactor was then purged with nitrogen and cooled. The final product (2.52 kg; 6.76 lb) contained 56.9 parts of poly(methyl methacrylate) (PMMA) per hundred parts of propylene homopolymer.

The polypropylene grafted with poly(methyl methacrylate-co-methyl acrylate) (PP-g-(MMA-co-MeAc) in Control 2 was prepared as follows. The propylene homopolymer used as the backbone polymer for the graft copolymer was the same as in Example 1. The monomers (95.6% methyl methacrylate, 4.4% methyl acrylate) were grafted onto the propylene homopolymer backbone at a grafting temperature of 114°C.

Ninety-five parts by weight monomers were added per 100 parts of propylene homopolymer (pph). Tert-butylperoxy-2-ethyl hexanoate (2.1 pph active peroxide, 50% in mineral spirits) was used as the peroxide initiator. The methyl methacrylate (MMA) and methyl acrylate were premixed and fed at 1.0 pph/min for 95 minutes. A monomer to initiator ratio of 120 was used. After addition of the monomers was complete, the temperature was raised to 140°C for an additional 60 to 120 minutes under a nitrogen purge until the amount of unreacted MMA in the product was <500 parts per million.

The graft copolymer/clay composites and the graft copolymers without the clay were mixed with the amounts of Engage 8150 elastomer or olefin polymer impact modifier, broad molecular weight distribution propylene homopolymer (BMWD PP), and stabilizer shown in Table 3 by dry mixing in a plastic bag. The compositions were then extruded on a Leistritz 30 mm twin screw extruder at 180-190°C, 300 RPM, and a feed rate of 5.6-7.5 kg (15-20 lb)/hr. The samples were then molded on a Boy molding machine at 230°C (450°F) to prepare the samples for physical testing.

The olefin polymer composition used as the impact modifier in Sample 1 and Control Example 1 comprised (a) 40% propylene homopolymer, (b) 15% of an ethylene/propylene copolymer insoluble in xylene and containing about 90% ethylene, and (c) 45% of an ethylene/propylene copolymer that is soluble in xylene at room temperature and contained 50-55% ethylene, commercially available from Montell USA Inc.

The broad molecular weight distribution polypropylene (BMWD PP) had a MFR of 1.1 g/10 min, xylene insolubles at room temperature of 97.8%, and  $M_w/M_n > 6$ , and is commercially available from Montell USA Inc.

The Engage 8150 ethylene/octene copolymer elastomer used in Sample 2 and Control 2 contained 25% polymerized octene monomer and is commercially available from DuPont-Dow Elastomers.

The stabilizer was the same as the one used in Example 1.

The composition of the samples and the results of the physical property testing are given in Table 3.

Table 3				
Sample	1	Control 1	2	Control 2
BMWD PP (%)	17.32	17.23	42.01	42.01
Polyolefin impact modifier (%)	43.22	43.29	0.0	0.0
Stabilizer (%)	0.2	0.2	0.2	0.2
Engage 8150 elastomer (%)	0.0	0.0	20.2	20.2
PP-g-St (45- pph St) (%)	39.46	39.48	0.0	0.0
PP-g-(MMA-co-MeAc) (95 pph total monomers) (%)	0.0	0.0	0.0	37.79
PP-g-PMMA (56.9 pph) (%)	0.0	0.0	37.79	0.0
PSt in product (%)	10.95	12.25	0.0	0.0
PMMA in product (%)	0.0	0.0	14.28	13.54
Inorganic content in product (%)	1.0	0.0	0.7	0.0
MFR (dg/min)	7.3	6.0	14.28	8.9
Notched Izod impact (ft.lb/in)	3.2	1.6	1.0	5.3
Tensile strength (psi)	3013	2913	3990	3330
Flex. Modulus (kpsi)	143	140	174	186

The data show that for the polystyrene-grafted system, the incorporation of clay improves the impact strength without any loss in modulus and tensile strength. For the PMMA-grafted system, however, tensile strength improves while modulus and impact strength decrease. The deterioration of the impact strength and the modulus are due to

the degradation of the PMMA during processing. The higher melt flow of the composite compared to the control is evidence of the degradation of the PMMA.

Other features, advantages and embodiments of the invention disclosed herein will be readily apparent to those exercising ordinary skill after reading the foregoing disclosures. In this regard, while specific embodiments of the invention have been described in considerable detail, variations and modifications of these embodiments can be effected without departing from the spirit and scope of the invention as described and claimed.



## Claims:

1. A composite material comprising:
  - (1) a graft copolymer comprising a backbone of a particulate propylene polymer material having a weight average diameter of 0.4-7 mm, a surface area of at least 0.1 m<sup>2</sup>/g, and a pore volume fraction of at least 0.07, and wherein more than 40% of the pores in the particle have a diameter greater than 1 micron, to which is graft polymerized at least one grafting monomer capable of being polymerized by free radicals, and
  - (2) a smectite clay having exchangeable cations that has been treated with at least one organic swelling agent, uniformly dispersed in the particulate propylene polymer material, wherein the total inorganic content of the composite material is 0.5% to 10%, based on the total weight of the composite material.
2. The composite of claim 1 wherein the propylene polymer material used as the backbone of the graft copolymer is selected from the group consisting of:
  - (1) a homopolymer of propylene having an isotactic index greater than 80;
  - (2) a copolymer of propylene and an olefin selected from the group consisting of ethylene and C<sub>4-10</sub> alpha-olefins, provided that when the olefin is ethylene, the maximum polymerized ethylene content is 10%, and when the olefin is a C<sub>4-10</sub> alpha-olefin, the maximum polymerized content thereof is 20% by weight, the copolymer having an isotactic index greater than 85;
  - (3) a terpolymer of propylene and two olefins selected from the group consisting of ethylene and C<sub>4-8</sub> alpha-olefins, provided that the maximum polymerized C<sub>4-8</sub> alpha-olefin content is 20% by weight, and, when ethylene is one of the olefins, the maximum polymerized ethylene content is 5% by weight, the terpolymer having an isotactic index greater than 85;
  - (4) an olefin polymer composition comprising:
    - (a) 10% to 60% by weight of a propylene homopolymer having an isotactic index greater than 80 or a copolymer of monomers selected from the group consisting of (i) propylene and ethylene, (ii) propylene, ethylene and a C<sub>4-8</sub> alpha-olefin, and (iii) propylene and a C<sub>4-8</sub> alpha-olefin, the copolymer having a polymerized propylene content of more than 85% by weight and an isotactic index greater than 85;

(b) 5% to 25% by weight of a copolymer of ethylene and propylene or a C<sub>4-8</sub> alpha-olefin that is insoluble in xylene at room temperature; and  
(c) 30% to 70% by weight of an elastomeric copolymer of monomers selected from the group consisting of (i) ethylene and propylene, (ii) ethylene, propylene, and a C<sub>4-8</sub> alpha-olefin, and (iii) ethylene and a C<sub>4-8</sub> alpha-olefin, the copolymer optionally containing 0.5% to 10% by weight of a polymerized diene and containing less than 70% by weight of polymerized ethylene and being soluble in xylene at room temperature and having an intrinsic viscosity, measured in decahydronaphthalene at 135°C, of 1.5 to 4.0 dl/g,

wherein the total amount of (b) and (c), based on the total olefin polymer composition, is 50% to 90%, the weight ratio of (b)/(c) is less than 0.4 and the composition is prepared by polymerization in at least two stages and has a flexural modulus of less than 150 MPa; and

(5) a thermoplastic olefin comprising:

(a) 10% to 60% of a propylene homopolymer having an isotactic index greater than 80, or a copolymer of monomers selected from the group consisting of (i) ethylene and propylene, (ii) ethylene, propylene and a C<sub>4-8</sub> alpha-olefin, and (iii) ethylene and a C<sub>4-8</sub> alpha-olefin, the copolymer having a polymerized propylene content greater than 85% and an isotactic index of greater than 85;

(b) 20% to 60% of an amorphous copolymer of monomers selected from the group consisting of (i) ethylene and propylene, (ii) ethylene, propylene, and a C<sub>4-8</sub> alpha-olefin, and (iii) ethylene and a C<sub>4-8</sub> alpha-olefin, the copolymer optionally containing 0.5% to 10% of a polymerized diene, and containing less than 70% polymerized ethylene and being soluble in xylene at room temperature; and

(c) 3% to 40% of a copolymer of ethylene and propylene or a C<sub>4-8</sub> alpha-olefin that is insoluble in xylene at room temperature,

wherein the thermoplastic olefin has a flexural modulus of greater than 150 but less than 1200 MPa.

3. The composite of claim 2 wherein the propylene polymer material is a propylene homopolymer.

4. The composite of claim 1 wherein the grafting monomer is styrene.

5. The composite of claim 1 wherein the grafting monomers are methyl methacrylate and methyl acrylate.

6. The composite of claim 1 wherein the smectite clay is montmorillonite having sodium exchangeable cations.

7. A process for making a composite material comprising, in a substantially non-oxidizing environment, graft polymerizing at least one liquid monomer capable of being polymerized by free radicals to a particulate propylene polymer material having a weight average diameter of 0.4-7 mm, a surface area of at least 0.1 m<sup>2</sup>/g, and a pore volume fraction of at least 0.07, and wherein more than 40% of the pores in the particle have a diameter greater than 1 micron, in the presence of (1) a smectite clay having exchangeable cations that has been treated with at least one organic swelling agent, and (2) 0.1 parts to 6 parts per hundred parts of the propylene polymer material of an organic free radical polymerization initiator, at a reaction temperature of 60°C to 125°C, the composite material having an inorganic content of 0.5% to 10%, based on the total weight of the composite material.

8. The process of claim 7 wherein the propylene polymer material used as the backbone of the graft copolymer is selected from the group consisting of:

- (1) a homopolymer of propylene having an isotactic index greater than 80;
- (2) a copolymer of propylene and an olefin selected from the group consisting of ethylene and C<sub>4-10</sub> alpha-olefins, provided that when the olefin is ethylene, the maximum polymerized ethylene content is 10%, and when the olefin is a C<sub>4-10</sub> alpha-olefin, the maximum polymerized content thereof is 20% by weight, the copolymer having an isotactic index greater than 85;
- (3) a terpolymer of propylene and two olefins selected from the group consisting of ethylene and C<sub>4-8</sub> alpha-olefins, provided that the maximum polymerized C<sub>4-8</sub> alpha-olefin content is 20% by weight, and, when ethylene is one of the olefins, the maximum polymerized ethylene content is 5% by weight, the terpolymer having an isotactic index greater than 85;
- (4) an olefin polymer composition comprising:

- 5 (a) 10% to 60% by weight of a propylene homopolymer having an isotactic index greater than 80 or a copolymer of monomers selected from the group consisting of (i) propylene and ethylene, (ii) propylene, ethylene and a C<sub>4-8</sub> alpha-olefin, and (iii) propylene and a C<sub>4-8</sub> alpha-olefin, the copolymer having a polymerized propylene content of more than 85% by weight and an isotactic index greater than 85;
- (b) 5% to 25% by weight of a copolymer of ethylene and propylene or a C<sub>4-8</sub> alpha-olefin that is insoluble in xylene at room temperature; and
- 10 (c) 30% to 70% by weight of an elastomeric copolymer of monomers selected from the group consisting of (i) ethylene and propylene, (ii) ethylene, propylene, and a C<sub>4-8</sub> alpha-olefin, and (iii) ethylene and a C<sub>4-8</sub> alpha-olefin, the copolymer optionally containing 0.5% to 10% by weight of a polymerized diene and containing less than 70% by weight of polymerized ethylene and being soluble in xylene at room temperature
- 15 and having an intrinsic viscosity, measured in decahydronaphthalene at 135°C, of 1.5 to 4.0 dl/g,

wherein the total amount of (b) and (c), based on the total olefin polymer composition, is 50% to 90%, the weight ratio of (b)/(c) is less than 0.4 and the composition is prepared by polymerization in at least two stages and has a

20 flexural modulus of less than 150 MPa; and

- (5) a thermoplastic olefin comprising:
- 25 (a) 10% to 60% of a propylene homopolymer having an isotactic index greater than 80, or a copolymer of monomers selected from the group consisting of (i) ethylene and propylene, (ii) ethylene, propylene and a C<sub>4-8</sub> alpha-olefin, and (iii) ethylene and a C<sub>4-8</sub> alpha-olefin, the copolymer having a polymerized propylene content greater than 85% and an isotactic index of greater than 85;
- 30 (b) 20% to 60% of an amorphous copolymer of monomers selected from the group consisting of (i) ethylene and propylene, (ii) ethylene, propylene, and a C<sub>4-8</sub> alpha-olefin, and (iii) ethylene and a C<sub>4-8</sub> alpha-olefin, the copolymer optionally containing 0.5% to 10% of a polymerized

diene, and containing less than 70% polymerized ethylene and being soluble in xylene at room temperature; and

(c) 3% to 40% of a copolymer of ethylene and propylene or a C<sub>4-8</sub> alpha-olefin that is insoluble in xylene at room temperature,

5 wherein the thermoplastic olefin has a flexural modulus of greater than 150 but less than 1200 MPa.

9. The process of claim 8 wherein the propylene polymer material is a propylene homopolymer.

10. The process of claim 7 wherein the grafting monomer is styrene.

10 11. The process of claim 7 wherein the grafting monomers are methyl methacrylate and methyl acrylate.

12. The process of claim 7 wherein the smectite clay is montmorillonite having sodium exchangeable cations.

13. The process of claim 7 wherein the particulate propylene polymer material is  
15 heated to the reaction temperature, the organic free radical initiator is added, and a suspension of the clay in liquid monomer is then added separately.

14. The process of claim 7 wherein the particulate propylene polymer material is mixed with a suspension of the clay in liquid monomer, the mixture is heated to the reaction temperature, and the organic free radical initiator is added.

20 15. The process of claim 7 wherein a suspension of the clay in liquid monomer and the organic free radical initiator are added concurrently to the particulate propylene polymer material, which has been heated to the reaction temperature.

16. The process of claim 7 wherein the clay is mixed with the particulate propylene polymer material in the presence of a hydrocarbon solvent, the mixture is  
25 heated to the reaction temperature to remove the solvent, and the organic free radical initiator and liquid monomer are added concurrently.

17. A composition comprising the composite material of claim 1 and 2% to 45% of a rubber component, based on the total weight of the composition.

18. A composition comprising the composite material of claim 1 and 5% to 90%  
30 of a broad molecular weight distribution propylene polymer material, having a  $M_w/M_n$  of 5 to 60.

19. The composition of claim 17, further comprising 5% to 90% of a broad molecular weight distribution propylene polymer material, having a  $M_w/M_n$  of 5 to 60.

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
20 December 2001 (20.12.2001)

PCT

(10) International Publication Number  
**WO 01/96467 A3**

- (51) International Patent Classification<sup>7</sup>: C08K 3/34, 9/04, C08F 255/02
- (21) International Application Number: PCT/IB01/01022
- (22) International Filing Date: 8 June 2001 (08.06.2001)
- (25) Filing Language: English
- (26) Publication Language: English
- (30) Priority Data: 09/596,092 16 June 2000 (16.06.2000) US
- (71) Applicant: BASELL TECHNOLOGY COMPANY B.V. [NL/NL]; Hoeksteen 66, NL-2132 MS Hoofddorp (NL).
- (72) Inventor: NIYOGI, Suhas, G.; 24 Clark Ridge, Hockessin, DE 19707 (US).
- (81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW.
- (84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).
- Declaration under Rule 4.17:  
— as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii)) for all designations
- Published:  
— with international search report
- (88) Date of publication of the international search report: 30 May 2002
- For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

WO 01/96467 A3

(54) Title: COMPOSITE MATERIALS COMPRISING PROPYLENE GRAFT COPOLYMERS

(57) Abstract: A composite material comprises a smectite clay having exchangeable cations that has been treated with at least one organic swelling agent, uniformly dispersed in a graft copolymer having a backbone of a porous propylene polymer material, to which is graft polymerized at least one grafting monomer capable of being polymerized by free radicals, wherein the total inorganic content of the composite material is 0.5 % to 10 %, based on the total weight of the composite material. The composite material is made by graft polymerizing at least one liquid monomer capable of being polymerized by free radicals onto a porous propylene polymer material, in a non-oxidizing environment, in the presence of a smectite clay and an organic free radical polymerization initiator, whereby the chains of polymerized monomer that are formed intercalate the clay and produce a uniform dispersion of clay particles within the particulate propylene polymer material.

## INTERNATIONAL SEARCH REPORT

International Application No

PCT/01/01022

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C08K3/34 C08K9/04 C08F255/02

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C08K C08F C08L

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 995 591 A (SHOWA DENKO KK) 26 April 2000 (2000-04-26) page 3, line 1-50 ---	1
A	EP 0 472 344 A (UBE INDUSTRIES ; TOYOTA MOTOR CO LTD (JP); TOYODA CHUO KENKYUSHO KK) 26 February 1992 (1992-02-26) claim 1 ---	1
E	WO 01 48080 A (FIBIGER RICHARD F ; CHOU CHAI JING (US); DOW CHEMICAL CO (US); SCHI) 5 July 2001 (2001-07-05) the whole document --- -/--	1

☒ Further documents are listed in the continuation of box C.☒ Patent family members are listed in annex.

## \* Special categories of cited documents:

\*A\* document defining the general state of the art which is not considered to be of particular relevance

\*E\* earlier document but published on or after the international filing date

\*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

\*O\* document referring to an oral disclosure, use, exhibition or other means

\*P\* document published prior to the international filing date but later than the priority date claimed

\*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

\*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

\*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

\*Z\* document member of the same patent family

Date of the actual completion of the international search

4 January 2002

Date of mailing of the international search report

17/01/2002

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl.  
Fax: (+31-70) 340-3016

Authorized officer

Meulemans, R



## INTERNATIONAL SEARCH REPORT

Inter. Appl. No.

PCT/01/01022

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	ALEXANDRE M ET AL: "Polymer-layered silicate nanocomposites: preparation, properties and uses of a new class of materials" MATERIALS SCIENCE AND ENGINEERING R: REPORTS, ELSEVIER SEQUOIA S.A., LAUSANNE, CH, vol. 28, no. 1-2, June 2000 (2000-06), pages 1-63, XP004205143 ISSN: 0927-796X the whole document ---	1
A	US 4 071 494 A (GAYLORD NORMAN G) 31 January 1978 (1978-01-31) column 2, line 39 -column 3, line 64 ---	1
A	US 4 810 734 A (KAWASUMI MASAYA ET AL) 7 March 1989 (1989-03-07) cited in the application claim 1 -----	1,7

# INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PC 01/01022

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
EP 0995591	A	26-04-2000	JP 2000129056 A EP 0995591 A1	09-05-2000 26-04-2000
EP 0472344	A	26-02-1992	JP 2885490 B2 JP 4096958 A JP 2885491 B2 JP 4096956 A JP 2885492 B2 JP 4096957 A JP 2885507 B2 JP 4183733 A EP 0472344 A2	26-04-1999 30-03-1992 26-04-1999 30-03-1992 26-04-1999 30-03-1992 26-04-1999 30-06-1992 26-02-1992
WO 0148080	A	05-07-2001	WO 0148080 A1	05-07-2001
US 4071494	A	31-01-1978	US 3956230 A CH 574464 A5 DE 2060665 A1 FR 2101334 A5 GB 1300640 A JP 49026700 B NL 7018507 A US 4317765 A	11-05-1976 15-04-1976 09-03-1972 31-03-1972 20-12-1972 11-07-1974 23-02-1972 02-03-1982
US 4810734	A	07-03-1989	JP 1009202 A JP 2098927 C JP 7078089 B DE 3810006 A1	12-01-1989 22-10-1996 23-08-1995 06-10-1988